

LECTURE # 14, Wednes., 11-2

*) Continue discussion from last time:

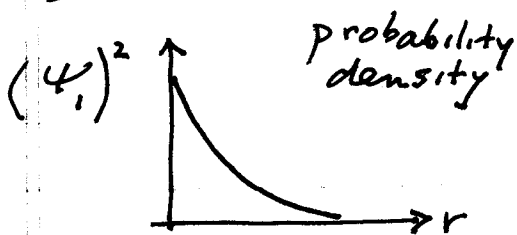
→ probability of finding electron ⁿ⁼¹ between r and $r+dr$ from nucleus, i.e., inside spherical shell] radius r thickness dr

$$= (\psi_1(r, \theta, \phi))^2 (\text{volume of shell})$$

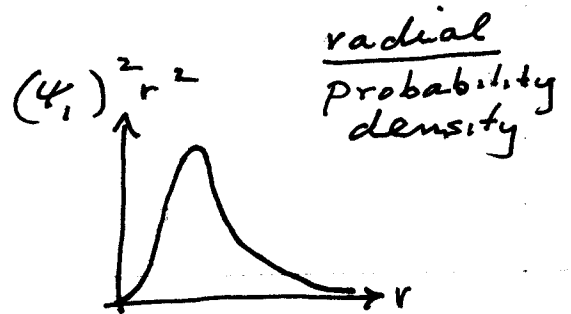
$$\frac{1}{\pi} \frac{1}{a_0^3} e^{-2r/a_0} \quad 4\pi r^2 dr$$

$$= \frac{4}{a_0^3} e^{-2r/a_0} r^2 dr$$

So:



whereas



*) Wavefunctions for higher-energy states are more complicated

For example

$$E_{n=2} = -\frac{R}{2^2} = -\frac{R}{4} \Rightarrow 4 \text{ different } \psi_2(r, \theta, \phi) \text{'s!}$$

$$\psi_{2s}(r, \theta, \phi) = \left(\frac{1}{4\pi}\right)^{1/2} \frac{1}{2\sqrt{2}} \left(\frac{1}{a_0}\right)^{3/2} \left(2 - \frac{r}{a_0}\right) e^{-r/2a_0}$$

$$\psi_{2p_x}(r, \theta, \phi) = \left(\frac{3}{4\pi}\right)^{1/2} \sin\theta \cos\phi \frac{1}{2\sqrt{6}} \left(\frac{1}{a_0}\right)^{3/2} \frac{r}{a_0} e^{-r/2a_0}$$

$$\psi_{2p_y}(r, \theta, \phi) = \left(\frac{3}{4\pi}\right)^{1/2} \sin\theta \sin\phi \quad "$$

$$\psi_{2p_z}(r, \theta, \phi) = \left(\frac{3}{4\pi}\right)^{1/2} \cos\theta \quad "$$

NOTE
MISTAKE
in Table 15.2

For $E_{n=3} = -\frac{R}{3^2} = -\frac{R}{9}$, things are even more complicated: instead of 1 allowed state with this energy (as for E_1), or 3 states (as for E_2), there are ... 9! (AND SO ON, FOR $n \geq 4$)

SEE Table 15.2, p. 538

→ Each allowed state corresponds, not only to a special energy (E_1 , or E_2 , or E_3 , ... $E_n = -R/n^2, n=1,2,\dots$) but also to a special (also quantized) angular "mv r" momentum: 0 , or $\sqrt{2} \frac{h}{2\pi}$, or $\sqrt{6} \frac{h}{2\pi}$, ... $\sqrt{l(l+1)} \frac{h}{2\pi}, l=0,1,2,\dots$

E.g., $\psi_1 \leftrightarrow l=0$

$\psi_{2s} \leftrightarrow l=0$

$\psi_{2p_x} \leftrightarrow l=1$
 ψ_{2p_y}
 ψ_{2p_z}

For $n=3$ states, there is one (ψ_{3s}) with $l=0$, 3 ($\psi_{3p_x, y, z}$) with $l=1$, and 5 ($\psi_{3d's}$)

with $l=2$

→ IN GENERAL, THERE ARE n^2 ALLOWED STATES CORRESPONDING TO THE ENERGY E_n

n^2	($E_{n=1}$)	$l=0$	$l=1$	$l=2$	$l=3$
1	($E_{n=1}$)	1s			
4	($E_{n=2}$)	2s	3 2p's		
9	($E_{n=3}$)	3s	3 3p's	5 3d's	
16	($E_{n=4}$)	4s	3 4p's	5 4d's	7 4f's